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International application number: PCT/US04/043887

International filing date: 30 December 2004 (30.12.2004)

Document type: Certified copy of priority document

Document details: Country/Office: US  
Number: 60/533,097  
Filing date: 30 December 2003 (30.12.2003)

Date of receipt at the International Bureau: 09 February 2005 (09.02.2005)

Remark: Priority document submitted or transmitted to the International Bureau in compliance with Rule 17.1(a) or (b)



World Intellectual Property Organization (WIPO) - Geneva, Switzerland  
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**APPLICATION NUMBER: 60/533,097**

**FILING DATE: *December 30, 2003***

**RELATED PCT APPLICATION NUMBER: *PCT/US04/43887***



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123003

17621 U.S. PTO

PTO/SB/16 (08-03)

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# PROVISIONAL APPLICATION FOR PATENT COVER SHEET

This is a request for filing a PROVISIONAL APPLICATION FOR PATENT under 37 CFR 1.53 (c).

Express Mail Label No. EV964554333US

031431 U.S. PTO  
60/533097

123003

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<input checked="" type="checkbox"/> Additional inventors are being named on the 1 separately numbered sheets attached hereto					
TITLE OF THE INVENTION (500 characters max)					
METHOD AND APPARATUS FOR HIGH SELECTIVITY SILICON NITRIDE ETCHING					
Direct all correspondence to: CORRESPONDENCE ADDRESS					
<input checked="" type="checkbox"/> Customer Number: 34132					
OR					
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ENCLOSED APPLICATION PARTS (check all that apply)					
<input checked="" type="checkbox"/> Specification Number of Pages 6 <input type="checkbox"/> CD(s), Number					
<input checked="" type="checkbox"/> Drawing(s) Number of Sheets 2 <input checked="" type="checkbox"/> Other (specify) Rotandaro, A.L.P., Hames, G.A., and Yocum, T. Electrochem. Soc. Proc. Vol. 99-36, pp. 385-390, which are incorporated by references into the specification					
<input type="checkbox"/> Application Data Sheet. See 37 CFR 1.76					
METHOD OF PAYMENT OF FILING FEES FOR THIS PROVISIONAL APPLICATION FOR PATENT					
<input checked="" type="checkbox"/> Applicant claims small entity status. See 37 CFR 1.27.					
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FILING FEE Amount (\$): 80.00					
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[Page 1 of 2]

Respectfully submitted, Brian L. Belles  
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REGISTRATION NO. 51,322  
(if appropriate)  
Docket Number: 108430.040

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Docket Number 108430.040

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**Patent Disclosure**  
***CONFIDENTIAL***

**METHOD AND APPARATUS FOR HIGH SELECTIVITY SILICON  
NITRIDE ETCHING**

**Ismail Kashkoush, Gim Chen and Rich Novak**

***December 21, 2003***



## **Patent Disclosure**

### **CONFIDENTIAL**

#### **OBJECTIVE**

To develop a process that yields a high selectivity silicon nitride ( $\text{Si}_3\text{N}_4$ ) to silicon oxide ( $\text{SiO}_2$ ) etch rates in IC manufacturing. The process of etching  $\text{Si}_3\text{N}_4$  in phos acid is commonly used in IC manufacturing because of its stable etch rate. However, the selectivity suffers i.e. increases over the bath life. The number of wafers processed in the bath influence the selectivity. Etch byproducts suppress the oxide etch rate.

#### **MATERIAL AND EXPERIMENTAL SETUP**

An acid bath was used to etch SN wafers. The bath is filtered, heated and filtered. Process conditions were: Sulfuric:Phos:water

Temp. 165 C

Rinse: 3 QDR + 10 min. cascade

Concentration control to maintain the mix ration throughout the bath

Feed and bleed may be required

- 200mm nitride wafers and thermal oxide wafers were used.
- System configured with recirculated PRM and  $\text{N}_2$  sparging
- Bath temperature = 165 deg. C
- Prior to testing, wafers were run through dilute HF for surface normalization.
- Test wafers from different sources were run together and showed slightly different etch rates. The average was used.
- Etch rate measurement was performed using Rudolph S300 with 49 point measurements in 5mm edge exclusion.
- Nitride loading effects were tested by processing full lots of nitride wafers for an extended period of time.

#### **Methods of Control**

- a. Concentration sensor e.g. NIR, FT-NIR: the system will monitor the concentration of sulfuric acid, phosphoric acid and water. The user will select the setpoint and the system will maintain the setpoint by injecting the right constituent to adjust. For example, the system will inject sulfuric acid if it goes lower than the specified value.
- b. Feed and bleed: to reduce the effect of the etch by products, a supply of chemicals will be activated. The user will program the frequency and volumes.

#### **RESULTS AND DISCUSSION**

Figure 1 shows the test apparatus where the acid mix flows from to process tank thru a concentration sensor, the pump, the heater, filters, and then back to the process tank.. The mix of chemicals is initially dispensed from the dispense lines according to the desired mix ratio. The process controller constantly monitors the chemicals concentrations and adjusts accordingly from the spike lines by opening and closing the control valves.

Figure 2 shows the etch rate (ER) of silicon nitride to silicon dioxide over time. As the data show, the selectivity obtained (ER of  $\text{Si}_3\text{N}_4$  / ER of  $\text{SiO}_2$ ) was higher than traditional nitride etch in phosphoric acid. Typical, a selectivity of ~ 40-50 is obtained. Here, a selectivity of > 200 was easily



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obtained. While these are very encouraging results but the control system is a critical factor to maintain the stability of the bath and selectivity over time. In IC manufacturing environment, the etch by-products saturate the process bath and suppress the  $\text{SiO}_2$  etch and thus yielding inconsistent etch over time i.e. lower device yield. By monitoring the acids concentrations and refreshing the bath with fresh acids (sulfuric and phosphoric), this ensures that the bath has a low etch byproducts and thus consistent etch rate over time.

### **CONCLUSION**

A process that utilizes a mix of sulfuric acid and phosphoric acid was developed. The process control also enabled us to extend the bath for many hours while achieving the same high ratio of silicon nitride to oxide etch rate.

### **REFERENCES:**

Rotandaro, A.L.P., Hames, G.A., and Yocum, T, Electrochem. Soc. Proc. Vol. 99-36, pp. 385-390.

What is claimed is:

1. A method of etching silicon nitride from at least one substrate comprising:
  - supplying sulfuric acid, phosphoric acid, and water to a process chamber in predetermined amounts so as to form a mixture having a predetermined concentration ratio;
  - overflowing the process chamber with the mixture;
  - recirculating overflowed mixture back into the process chamber via a recirculation line;
  - submerging at least one substrate in the mixture within the process chamber;
  - constantly monitoring the concentration ratio of the mixture with a concentration sensor;
  - upon the concentration sensor detecting the mixture having a concentration ratio other than the predetermined concentration ratio, automatically adding an amount of sulfuric acid, phosphoric acid, and/or water to the mixture to return the concentration ratio of the mixture back to or near the predetermined concentration ratio.
2. The method of claim 1 wherein the sulfuric acid, phosphoric acid, and water are initially supplied to the process chamber via dispense lines.
3. The method of claim 1 wherein the amount of sulfuric acid, phosphoric acid, and/or water added to the mixture to return the concentration ratio of the mixture to or near the predetermined concentration ratio are supplied via spike lines.
4. The method of claim 1 further comprising heating the mixture prior to submerging the at least one wafer therein.
5. The method of claim 4 wherein the mixture is heated to a temperature at or near 160-180°C.
- 5A. The method of claim 5 wherein the mixture is heated to a temperature at or near 165°C.
6. The method of claim 1 wherein the predetermined concentration ratio is approximately 2 parts sulfuric acid, 2 parts phosphoric acid, and 1 part water.
7. The method of claim 1 wherein the concentration sensor measures the concentration ratio of the mixture along a recirculation line.
8. The method of claim 1 further comprising filtering the overflowed mixture.



9. The method of claim 1 further comprising:

bleeding a volume of mixture so as to reduce the effect of etch by products; and  
adding phosphoric acid, sulfuric acid, and/or water to replace the volume of  
mixture bled.

10. The method of claim 9 wherein a total volume of phosphoric acid, sulfuric acid,  
and/or water added to replace the bled mixture is approximately equal to the volume of  
the mixture drained.

11. The method of claim 9 wherein the mixture is bled at set intervals or continuously.

12. The method of claim 11 wherein the bleeding can be done while at least one  
substrate is submerged in the mixture or between batches of substrates.

13. A method of etching silicon nitride from at least one substrate comprising:

supplying sulfuric acid, phosphoric acid, and water to a process chamber in  
predetermined amounts so as to form a predetermined volume of mixture having a  
predetermined concentration ratio;

circulating the mixture through the process chamber in a closed-loop system;  
submerging at least one substrate in the mixture within the process chamber;  
constantly monitoring the concentration ratio of the mixture with a concentration  
sensor;

bleeding a volume of mixture from the closed loop system so as to reduce the  
effect of etch by products in the circulating mixture; and

adding phosphoric acid, sulfuric acid, and/or water to replace the volume of  
mixture bled from the closed loop in amounts to maintain or return the concentration ratio  
of the mixture back to the predetermined concentration ratio.

14. The method of claim 13 wherein the bleeding is continuous or at set intervals.

15. The method of claim 13 further comprising upon the concentration sensor detecting  
the mixture having a concentration ratio other than the predetermined concentration ratio,  
automatically adding an amount of sulfuric acid, phosphoric acid, and/or water to the  
mixture to return the concentration ratio of the mixture back to or near the predetermined  
concentration ratio.

16. A system for etching silicon nitride from at least one substrate comprising:

a process chamber;

means to supply sulfuric acid, phosphoric acid, and water to the process chamber in amounts so as to form a mixture having a predetermined concentration ratio;

a recirculation line for circulating the mixture through the process chamber;

means for adding sulfuric acid, phosphoric acid, and water to the mixture as necessary;

means to control fluid flow through the spike lines;

a concentration sensor that continuously measures the concentration ratio of the mixture and produces a signal indicative of the measured concentration ratio; and

a processor coupled to the concentration sensor for receiving the signal, the processor programmed so that upon receiving a signal indicative of a concentration ratio other than the predetermined concentration ratio, the processor automatically adds an amount of sulfuric acid, phosphoric acid, and/or water to the mixture necessary to return the concentration ratio of the mixture back to or near the predetermined concentration ratio.

17. The system of claim 16 wherein the process chamber comprises an overflow weir.

18. The system of claim 16 wherein the process chamber is adapted to receive a plurality of substrates.

19. The system of claim 16 further comprising a filter operably coupled to the recirculation line.

20. The system of claim 16 further comprising a heater coupled to the recirculation line and adapted to heat the mixture.

21. The system of claim 16 wherein the means to supply sulfuric acid, phosphoric acid, and water to the process chamber are a plurality of dispense lines with valves.

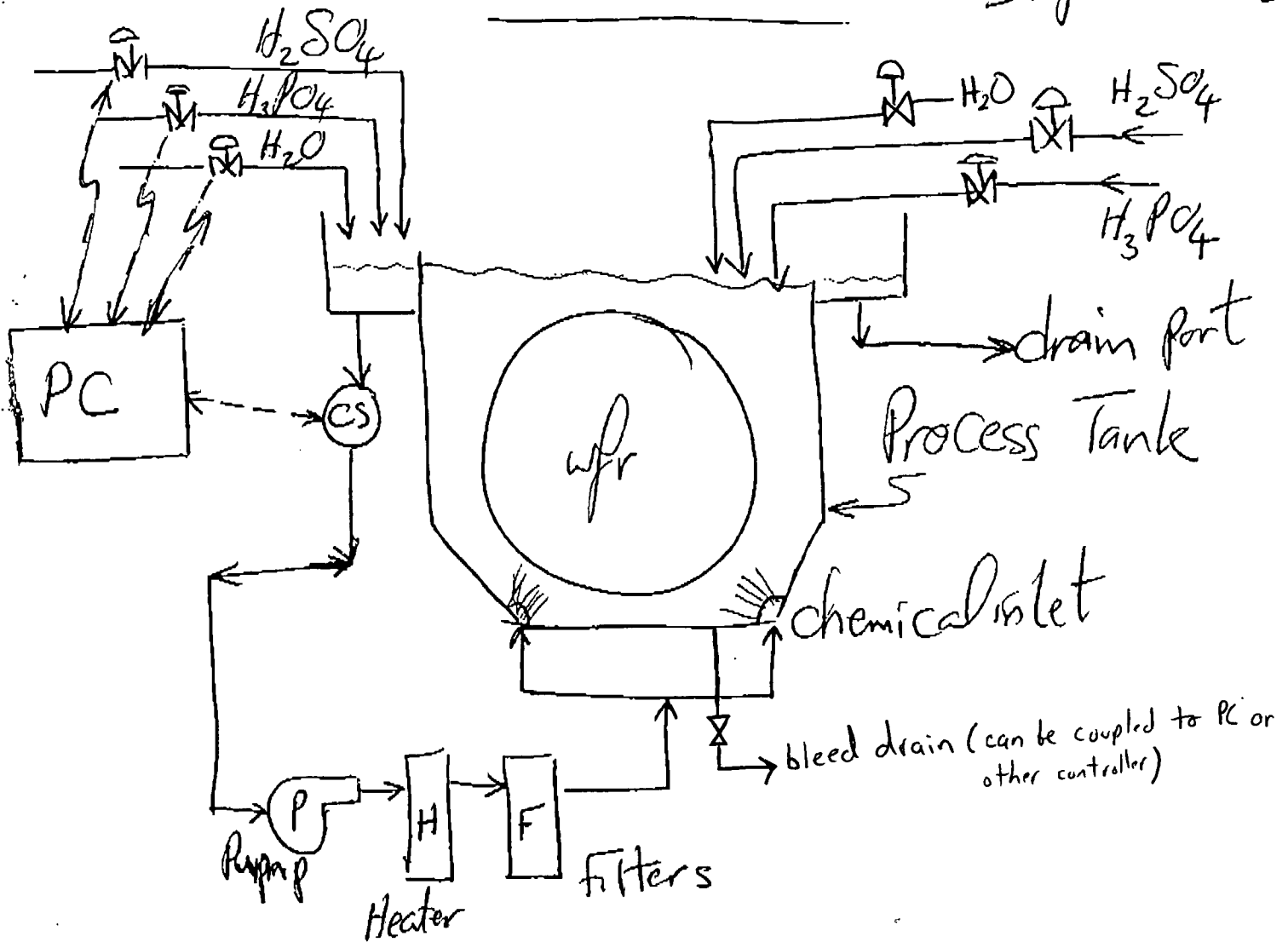
22. The system of claim 21 wherein the means to add sulfuric acid, phosphoric acid, and water to the mixture is a plurality of spike lines with valves.

23. The system of claim 16 further comprising a bleed drain for bleeding a volume of the mixture from the process chamber to reduce the effect of etch by products in the mixture.

Chemical  
Spike Lines

# PATENT DISCLOSURE

Chemical  
Dispense Lines



CS : Chemical Concentration Control Sensor  
 PC : Process Controller

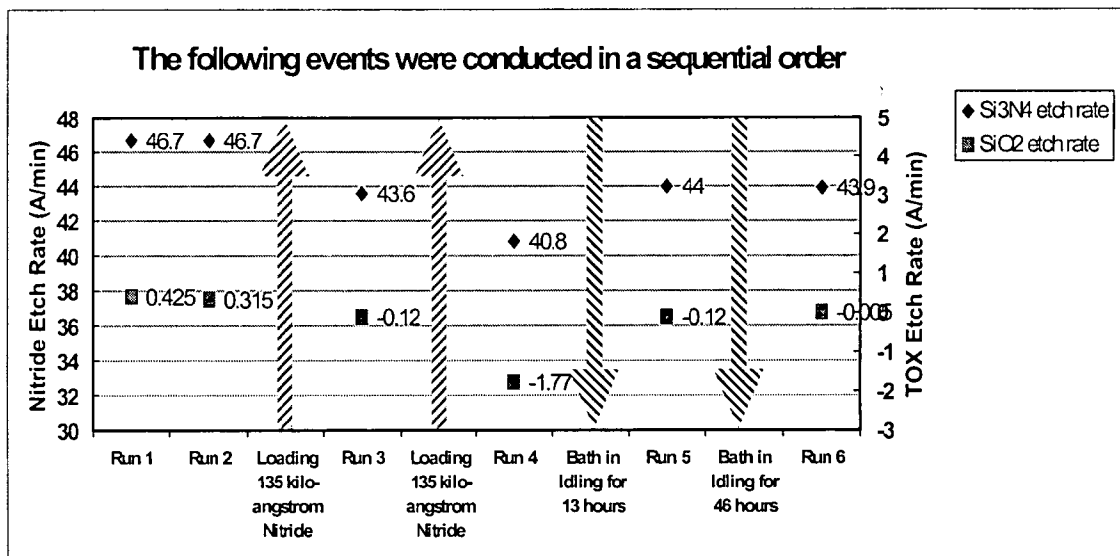
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Figure 1: Experimental Setup



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Figure 2: Silicon Nitride to Silicon Dioxide etch over time.



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